

**Cactus-like NC/Co<sub>x</sub>P electrode enables efficient and stable hydrogen evolution  
for saline water splitting**

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NEW CARBON MATERIALS

## Electrochemical Characterization

The corresponding Tafel plots were obtained by fitting the polarization curves between the overpotential and log current density by the equation:

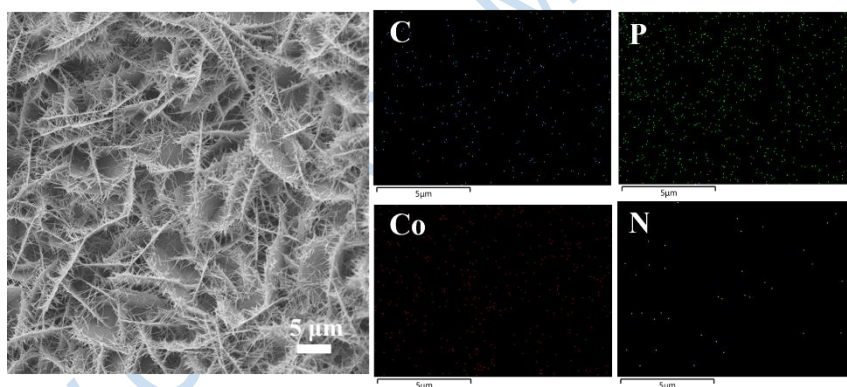
$$\eta = b \log(j) + a \quad (1)$$

where  $\eta$  is the overpotential,  $b$  is the Tafel slope,  $j$  is the current density.

Electrochemical active surface area (ECSA) were tested in the non-Faraday region, which could be calculated via the following equation:

$$ECSA = C_{dl}/C_s \quad (2)$$

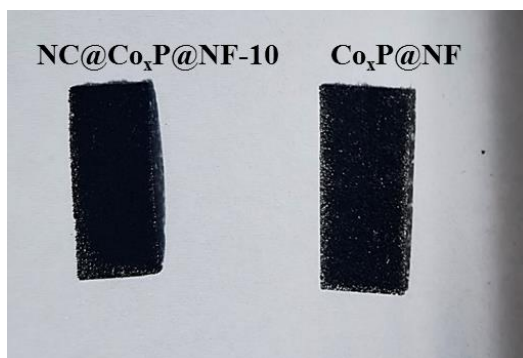
in which  $C_s$  is the specific capacitance for a flat surface and taken as  $0.040 \text{ mF cm}^{-2}$  in 1 M KOH.



**Figure S1.** SEM and the corresponding EDS elemental mapping images of the NC@Co<sub>x</sub>P@NF-10 catalyst.

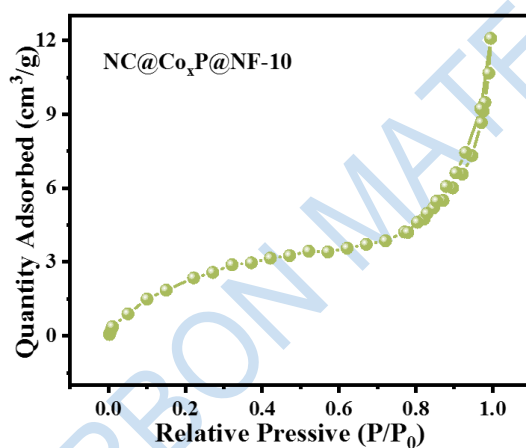
**Table S1.** The element content of the NC@Co<sub>x</sub>P@NF-10 sample.

Elements	wt%
C	8.90
N	1.42
P	6.08
Co	83.60

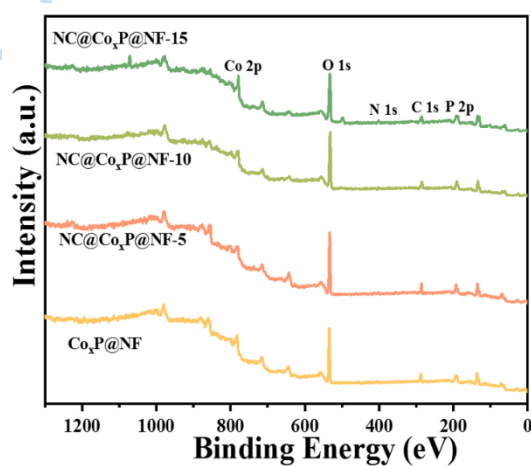


**Figure S2.** Images of NC@Co<sub>x</sub>P@NF-10 and Co<sub>x</sub>P@NF

after 30 min sonication.



**Figure S3.** N<sub>2</sub> adsorption/desorption isotherm of NC@Co<sub>x</sub>P@NF-10.

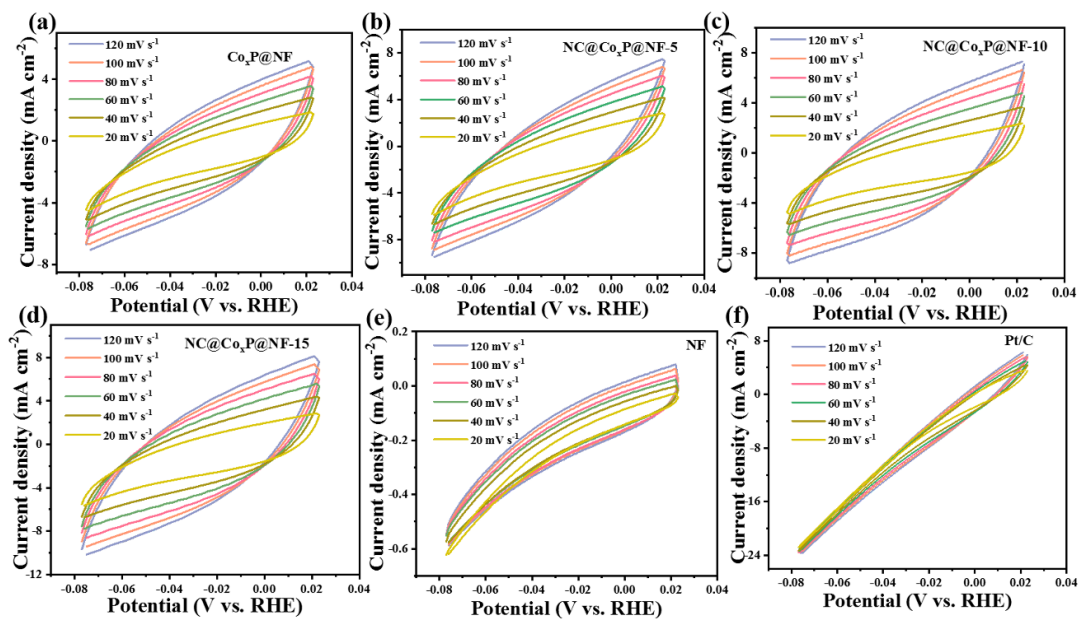


**Figure S4.** XPS survey spectra of Co<sub>x</sub>P@NF, NC@Co<sub>x</sub>P@NF-5, NC@Co<sub>x</sub>P@NF-10 and

NC@Co<sub>x</sub>P@NF-15 catalysts.

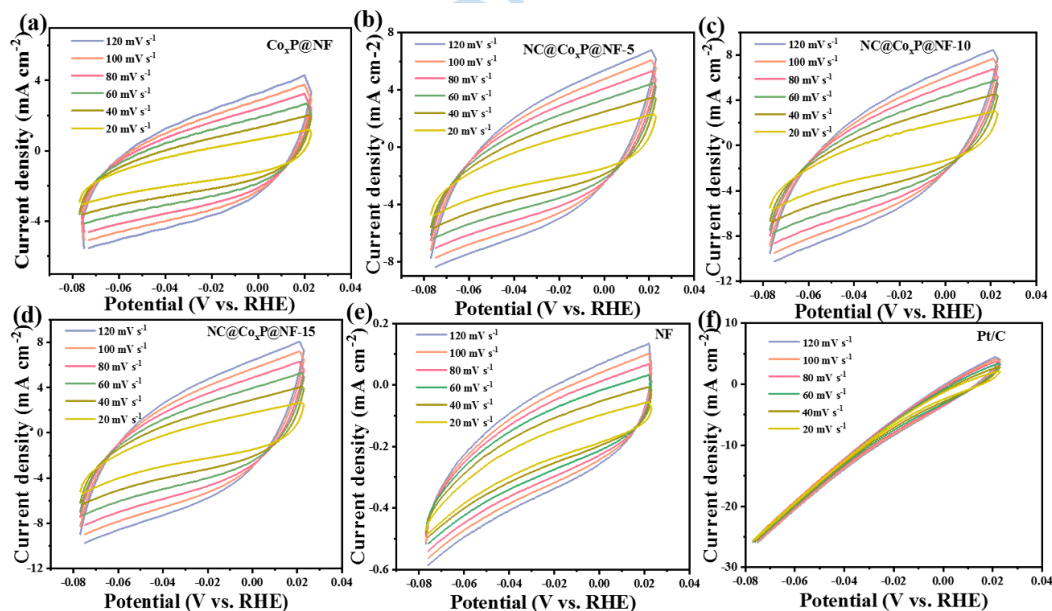
**Table S2.** Comparison of HER performance of NC@Co<sub>x</sub>P@NF-10 with other reported electrocatalysts at room temperature.

Catalysts	Overpotential@10 mA cm <sup>-2</sup>	Electrolyte	References
	107	1 M KOH	
NC@Co <sub>x</sub> P@NF-10	133	1 M KOH +0.5 M NaCl	This work
0.5CDs-NiCoP/NF	115	1 M KOH	[1]
IrCo <sub>3</sub> O <sub>4</sub> @NC	188	1 M KOH	[2]
Co(OH) <sub>2</sub> /Fe <sub>7</sub> Se <sub>8</sub>	183	1 M KOH	[3]
CoNC-SA/N*-C	194	1 M KOH	[4]
V <sub>2</sub> CT <sub>x</sub>	134	1 M KOH	[5]
SnS <sub>2</sub> /SnO <sub>2</sub>	108	1 M KOH	[6]
MnFeCrLDH	154	1 M KOH	[7]
	128	1 M KOH	
Ni <sub>2</sub> P-Fe <sub>2</sub> P/NF	155	1 M KOH +0.5 M NaCl	[8]
CdFe-BDC	148	1 M KOH	[9]
Ni-Co@Fe-Co PBA	183	1.0 M KOH + 0.5 M NaCl	[10]
NiFeP/SG	115	1 M KOH	[11]
NiFe(dobpdc)	113	1 M KOH	[12]
2S-MnCo <sub>2</sub> O <sub>4</sub> /NC	180	1 M KOH	[13]
CoP <sub>x</sub> @FeOOH/NF	117	1 M KOH	[14]
CoP/CoS <sub>2</sub> /CC	111.2	1 M KOH	[15]
	126	1 M KOH	
P-Fe <sub>2</sub> O <sub>3</sub> -CoP	152	1 M KOH+0.5 M NaCl	[16]



**Figure S5.** CV curves at different scan rate for (a)  $\text{Co}_x\text{P@NF}$ , (b)  $\text{NC@Co}_x\text{P@NF-5}$ , (c)

$\text{NC@Co}_x\text{P@NF-10}$ , (d)  $\text{NC@Co}_x\text{P@NF-15}$ , (e) NF and (f) Pt/C in 1 M KOH.



**Figure S6.** CV curves at different scan rate for (a)  $\text{Co}_x\text{P@NF}$ , (b)  $\text{NC@Co}_x\text{P@NF-5}$ , (c)

$\text{NC@Co}_x\text{P@NF-10}$ , (d)  $\text{NC@Co}_x\text{P@NF-15}$ , (e) NF and (f) Pt/C in 1 M KOH + 0.5 M NaCl.

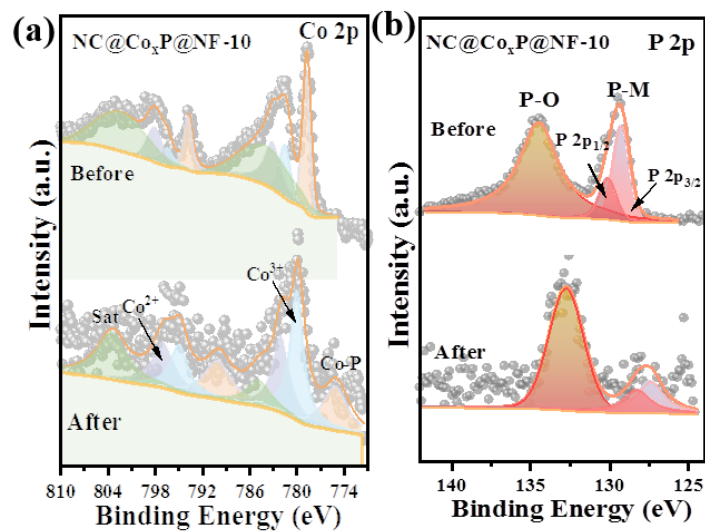


Figure S7. (a) Co 2p and (b) P 2p XPS spectra comparison of NC@Co<sub>x</sub>P@NF-10 before and after

HER test.

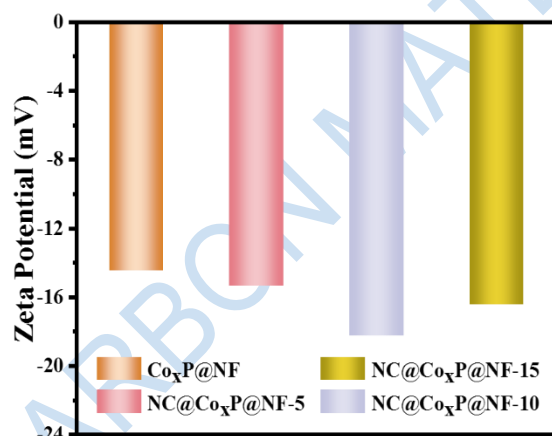


Figure S8. Zeta potential of all catalysts.

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