

Supporting Information

**Encapsulation of sulfur inside micro-nano carbon/molybdenum
carbide by in-situ chemical transformation for
high-performance Li-S batteries**

CHEN Xin-rong, YU Xiao-fei, HE Bin*, LI Wen-cui*

(State Key Laboratory of Fine Chemicals, School of Chemical Engineering, Dalian University of Technology, Dalian 116024, China)

Corresponding author: HE Bin, Associate Professor. E-mail: hebin71@dlut.edu.cn;

LI Wen-Cui, Ph.D, Professor. E-mail: wencuili@dlut.edu.cn

Author introduction: CHEN Xin-Rong, Master Student. E-mail: cxr170623@163.com

NEW CARBON MATERIALS

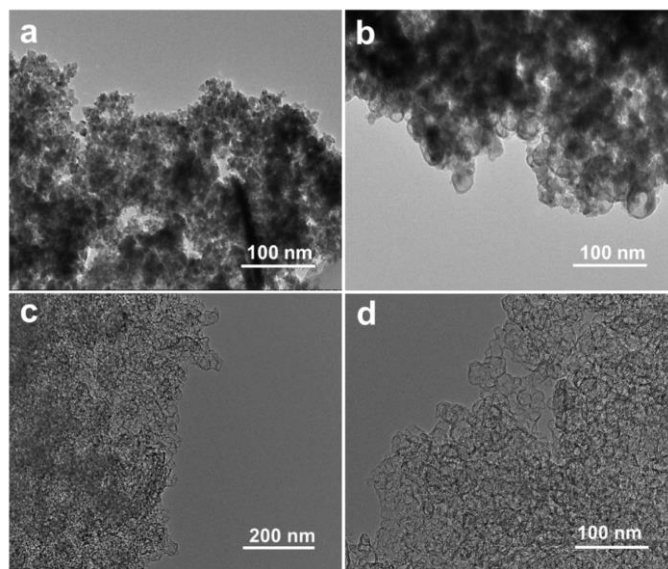


Fig. S1 TEM images of (a) ZnS, (b) ZnS@C/Mo₂C composite and (c,d) S@MC composite.

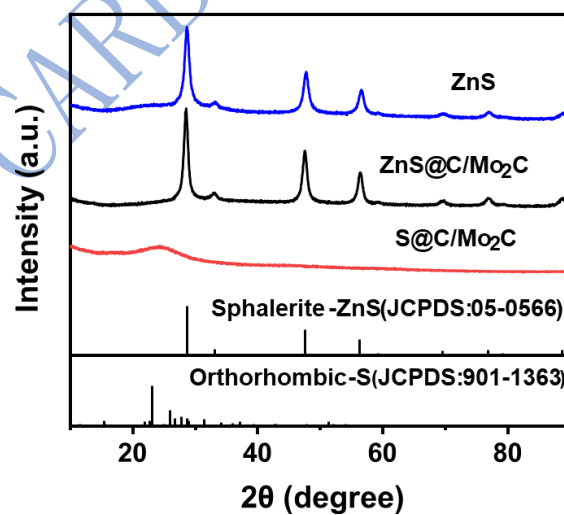


Fig. S2 XRD patterns of ZnS, ZnS@C/Mo₂C and S@C/Mo₂C.

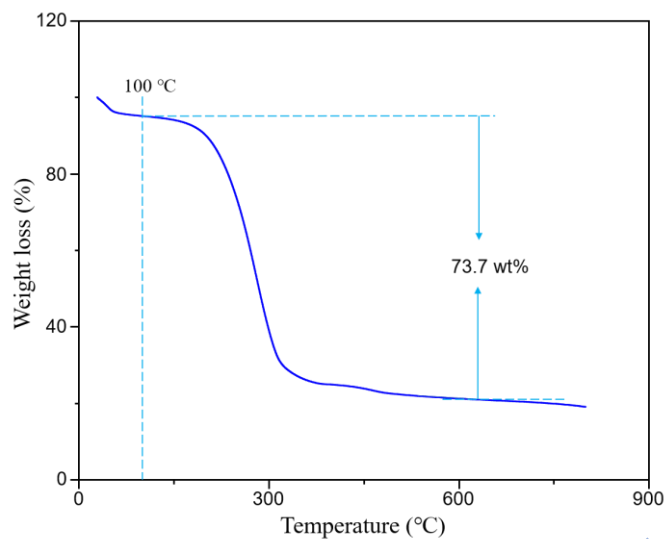


Fig. S3 Thermogravimetric analysis curves of S@C/Mo₂C in the nitrogen atmosphere.

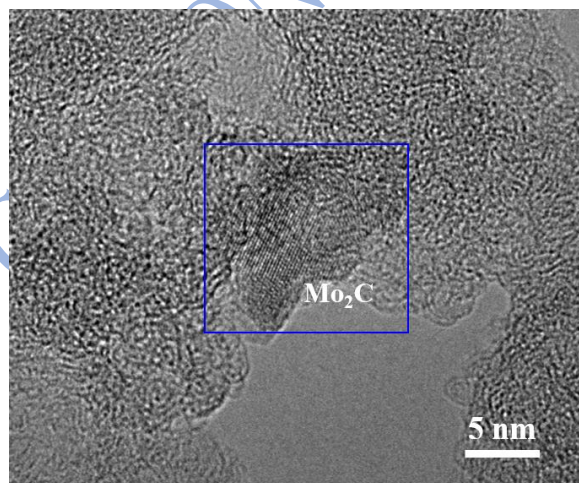


Fig. S4 High resolution TEM image of C/Mo₂C.

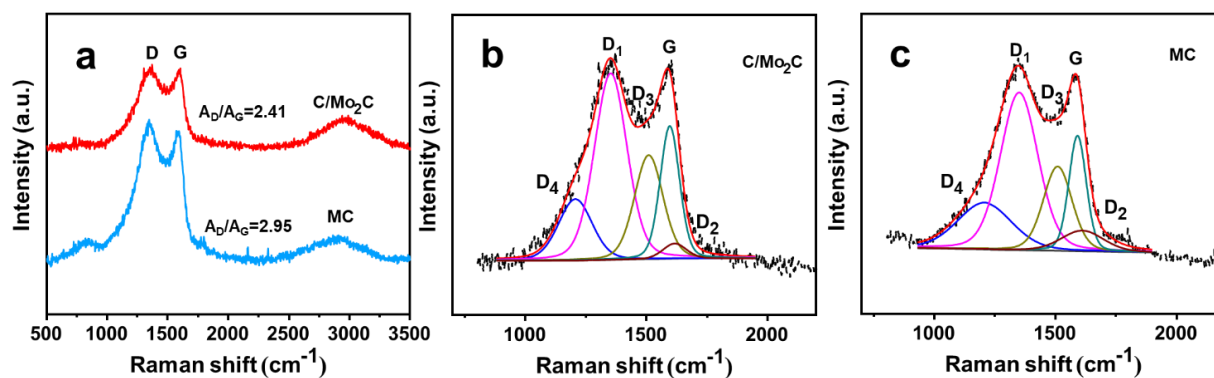


Fig. S5 (a) Raman spectra of the C/Mo₂C and MC. Curve fitting with band combination for the Raman spectra of (b) C/Mo₂C, (c) MC.

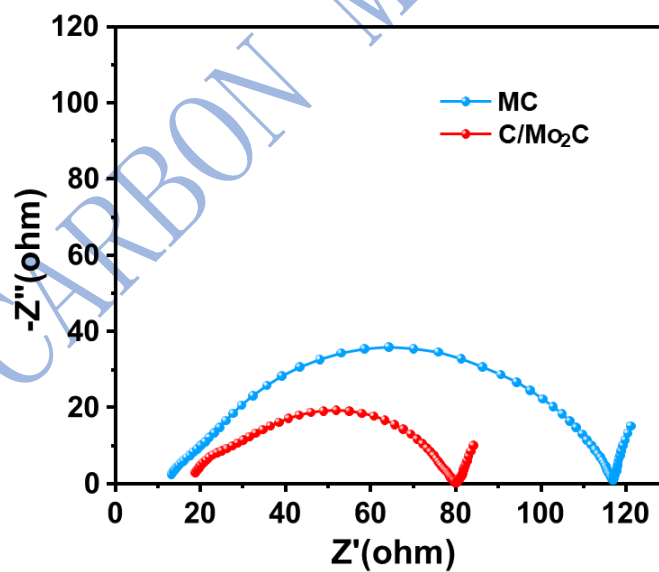


Fig. S6 EIS spectra of symmetric cells for C/Mo₂C and MC.

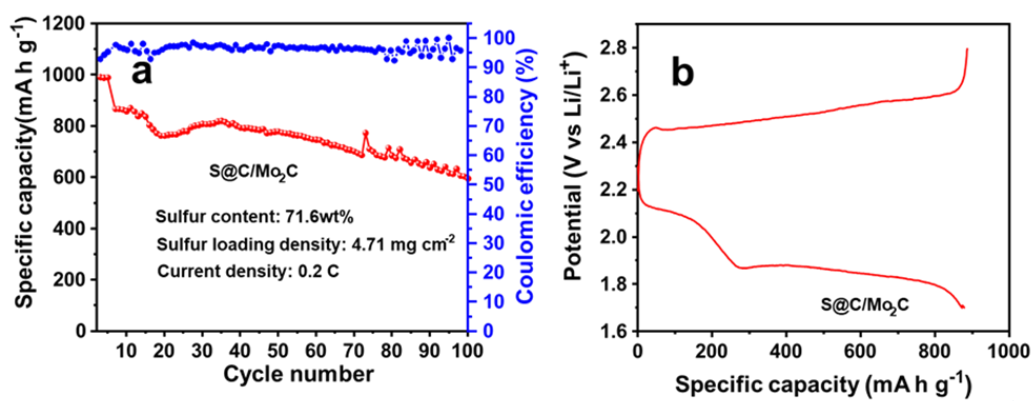


Fig. S7 (a) Cycling performance, (b) discharge/charge voltage profiles of S@C/Mo₂C with the sulfur loading of 4.71 mg cm⁻² at 0.2 C.

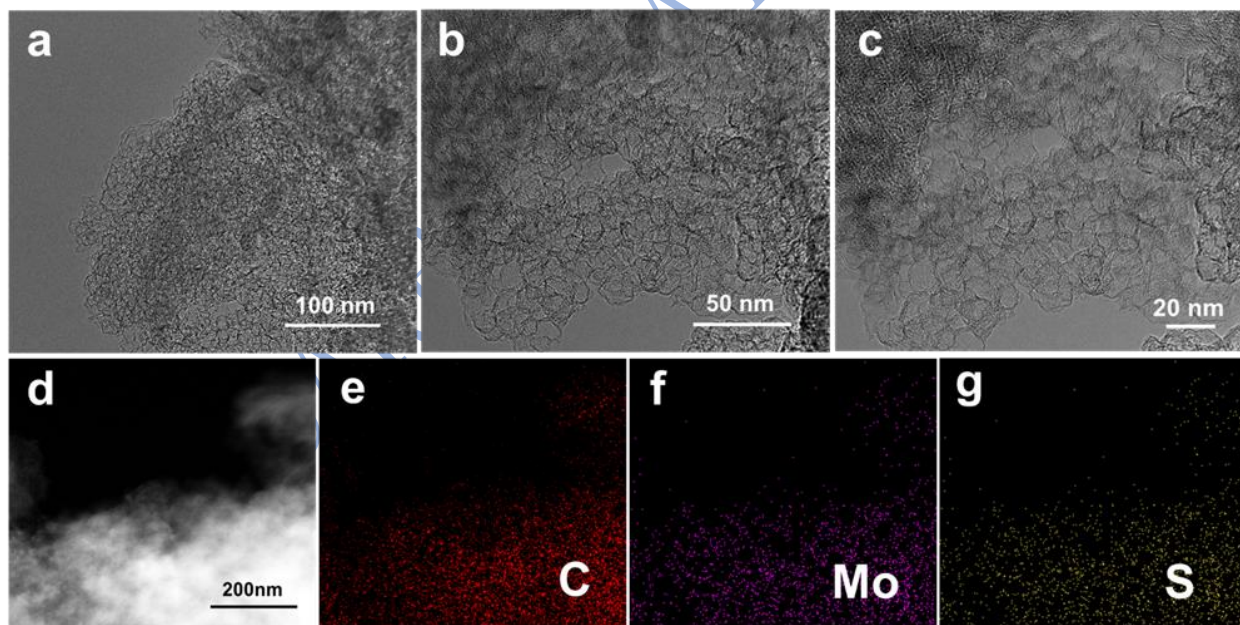


Fig. S8 (a-c) TEM images (d) STEM, EDS element mappings of (h) sulfur, (i) molybdenum, and (j) carbon of the S@C/Mo₂C composite after 60 cycles.